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# Atmospheric pressure plasma jet for bacterial decontamination and property improvement of fruit and vegetable processing wastewater

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## Abstract

An atmospheric pressure plasma jet was tested for decontaminating and improving the characteristics of wastewater derived from blackberry, date palm, tomato and beetroot processing industries. The jet was generated by blowing argon gas through a cylindrical alumina tube while a high voltage was applied between two electrodes surrounding the tube. Oxygen gas was mixed with argon at the rate of 0.2% and the argon mass flow was fixed at 4.5 slm. Images show that the generated plasma jet penetrated the treated wastewater samples. Plasma emission spectra show the presence of O and OH radicals as well as excited molecular nitrogen and argon. Complete decontamination of wastewater derived from date palm and tomato processing was achieved after 120 and 150 s exposure to the plasma jet, respectively. The bacterial count of wastewater from blackberry and beetroot was reduced by 0.41 and 2.24 log<sub>10</sub> colony-forming units (CFU) per ml, respectively, after 180 s. *Escherichia coli* was the most susceptible bacterial species to the cold plasma while *Shigella boydii* had the minimum susceptibility, recording 1.30 and 3.34 log<sub>10</sub> CFU ml<sup>-1</sup>, respectively, as compared to the 7.00 log<sub>10</sub> initial count. The chemical oxygen demands of wastewater were improved by 57.5–93.3% after 180 s exposure to the plasma jet being tested. The endotoxins in the wastewater were reduced by up to 90.22%. The variation in plasma effectiveness is probably related to the antioxidant concentration of the different investigated wastewaters.

Keywords: cold plasma jet, wastewater, bacterial decontamination, antioxidants, endotoxins

(Some figures may appear in colour only in the online journal)

## 1. Introduction

The food-processing industry consumes water to meet its individual daily needs. Most of the water is used in the fruit and vegetable sector for washing and rinsing. During these processes, water is contaminated with a high load of saprophytic and pathogenic organisms as well as hazardous and persistent compounds (Shaw and Schudram 2000). Wastewaters with high quantitative and qualitative pathogenic levels must be decontaminated prior to discharge. Typically, chlorine is used to decontaminate the wastewaters. Ultraviolet (UV) radiation, ozone, and other non-traditional decontamination processes have achieved some acceptable results due to stricter rules on the amount of residual chlorine levels in discharged wastewaters. However, industries are in search of disinfectants that are more efficacious against emerging and common pathogens as well as safe usage in many particular applications of food processing. A non-thermal plasma could represent a potential alternative technique for the disinfection of wastewater compared to the conventional methods.

Plasma is an ionized gas containing ions, electrons and reactive species such as atoms, molecules and radicals (OH, O, NO) as well as energetic photons (e.g. UV) (Moisan *et al* 2001). Plasma has been shown to be a sterilizing agent to inactivate a wide range of microorganisms (Deng *et al* 2005, Laroussi *et al* 2006, Kolb *et al* 2008, Park *et al* 2012, Hao *et al* 2014, El-Sayed *et al* 2015, Ouf *et al* 2015, 2016). Several researchers have demonstrated the application of non-thermal plasma in the food industry. Representative contributions include the application of plasma in the decontamination of grain and legumes infected by fungi (Selcuk *et al* 2008), inactivation of surface-borne microorganisms of *cicer arietinum* seeds (Mitra *et al* 2014), disinfection of packaged liquid food products (Misra *et al* 2012), treatment of ready-to-eat meat (Rød *et al* 2012), decontamination of almonds infected by *Escherichia coli* (Deng *et al* 2005), treatment of contaminated fresh fruit and vegetable slices (Wang *et al* 2012), and decontamination of the fruit pericarps from spoilage and pathogenic microorganisms (Perni *et al* 2008, Mohamed *et al* 2014). Many types of cold atmospheric pressure plasma sources have been established to provide reasonable reactivity with a low gas temperature. These plasmas are easy to set up because they do not require a vacuum system and are capable of running with numerous feeding gases under a wide range of driving powers and frequencies. However, the plasma operating conditions have to be optimized to induce maximum inactivation of the microorganisms and at the same time minimize the damage to the exposed materials (Anderson 1989, Lerouge *et al* 2000).

Atmospheric pressure plasma is not expensive compared to low pressure plasmas used for food processing due to the elimination of a vacuum system. It is feasible for use in food decontamination and water purification since it has the advantage of utilizing non-toxic gas and has no dangerous radiation. The reactive chemical species that are the main plasma antimicrobial effective agents are generated from the operating gas (such as He, N<sub>2</sub>, Ar, O<sub>2</sub>, Ne) and even from air. Atmospheric

pressure plasma is accessible and flexible enough to be used for different shapes, varieties of environments and in continuous decontamination chain processes.

The objective of the present study is to evaluate the possibility of using atmospheric pressure plasma to improve water quality by reducing the chemical oxygen demands and to decontaminate the wastewater of its bacterial load.

## 2. Materials and method

### 2.1. Sample collection

Thirty-two samples of wastewater were collected from three sites, Jeddah, Yanbu and Al-Madinah Al-Munawwarah, Saudi Arabia, during May 2012. The samples were collected from plant processing industries of blackberry, date palm, tomato and beetroot. The wastewater samples were collected in pre-sterilized 500 ml amber polypropylene high-density bottles equipped with Teflon-lined polypropylene caps. The samples were transferred into the laboratory immediately after collection to avoid unpredictable changes, then they were stored in a refrigerator at 2 °C until analyzed. The temperature recorded during the sample collection months ranged between 32 and 38 °C. Control samples of water before being used in processing did not contain more than 10<sup>3</sup> CFU ml<sup>-1</sup> bacteria.

### 2.2. Physical and chemical characteristics

**2.2.1. pH.** The pH value of each sample of wastewater, before and after cold plasma treatment, was determined using a pH meter (Ion 510, Bench pH/Ion/mV Meter, Oakton, USA) set directly into each solution. The pH meter, accurate to 0.1, was first calibrated according to the manufacturer's instructions, using buffer standards of pH 4 and pH 7. Twenty milliliters of each wastewater sample were placed in a 150 ml beaker, the pH meter electrode was immersed in the wastewater and the reading was recorded (Cavalcanti *et al* 2008). Inbetween readings, the electrode was washed in distilled water to make sure that no cross-contamination occurred. Each analysis was made in triplicate.

**2.2.2. Total suspended solids (TSS).** The TSS concentration was measured using a Hach 2100P Turbidimeter (Hach Company, Loveland, CO). The method is based on the theory that light scattering increases with the concentration of particles (Sadar 1998). The data were reported in NTU (nephelometric turbidity units).

**2.2.3. Total sugars and proteins.** The total proteins in the wastewater samples were determined using the method of Lowry *et al* (Lowry *et al* 1951). Standards of bovine serum albumin ranging in concentration from 0.01 to 1.0 mg ml<sup>-1</sup> were running simultaneously with the examined samples. The content of total sugars was estimated based on the phenol-sulphuric acid reaction of sugars (Saha and Brewer 1994). Glucose standards ranging in concentration from 0.01 to 0.1 mg ml<sup>-1</sup> were also tracked using the same method.

**2.2.4. Chemical oxygen demands (COD).** The examined sample was oxidized under reflux using a certain amount of potassium dichromate in strong sulphuric acid with silver sulphate as a catalyst. Organic matter reduced part of the dichromate and the residual part was estimated by titration with iron (II) ammonium sulphate using ferroin as indicator. Interference from chloride was inhibited by the addition of mercuric sulphate to the reaction mixture. The chemical COD was expressed in mg of oxygen absorbed from standard dichromate per liter of sample (APHA 1992).

**2.2.5. Antioxidant determination.** The antioxidants were assayed according to the method of ferric reducing antioxidant power (FRAP) described by Benzie and Strain (1999) using an Agilent 8453 Diode Array UV/VIS spectrophotometer, Model G1103A. The method is based on the reduction of the  $\text{Fe}^{3+}$ -TPTZ complex to the ferrous form at low pH. This reduction is measured using the absorption change at 593 nm. To conduct the assay, a 2.95 ml aliquot of a FRAP reagent, a mixture of  $0.1 \text{ mol l}^{-1}$  acetate buffer,  $10 \text{ mmol l}^{-1}$  2,4,6-tripyridyls-triazine (TPTZ), and  $20 \text{ mmol l}^{-1}$  ferric chloride (10:1:1 v/v/v), were combined with  $50 \mu\text{l}$  of the sample. The absorbance was recorded after 30 min incubation at  $37^\circ\text{C}$ . FRAP values were obtained by comparing the absorption change in the test mixture with those obtained from increasing concentrations of  $\text{Fe}^{3+}$  and expressed as mmol of  $\text{Fe}^{2+}$  equivalents per liter of sample.

### 2.3. Bacterial count

A bacterial count in the wastewater was performed, before and after plasma treatment, using the spread plate method. The exposure times of the cold plasma were 30, 60, 90, 120 and 150 s. To count the total number of CFU, 0.1 ml of wastewater samples diluted according to the expected number of colonies was spread on nutrient agar (Oxoid, UK) medium plates. The plates were incubated at  $28^\circ\text{C}$ . The total number of bacterial colonies were counted after 48 h of incubation. Measurements were carried out in triplicate.

A preliminary examination of the investigated wastewater using different specific media showed that the most common bacterial species were *E. coli*, *Pseudomonas aeruginosa*, *Enterococcus faecalis*, *Campylobacter jejuni* and *Shigella boydii*. The specific solid culture media employed for bacterial recognition were violet red bile lactose agar (Oxoid) for *E. coli*, cetrinide agar (Oxoid) for *P. aeruginosa*, bile esulin azide agar (Difco) for *E. faecalis*, campylobacter blood free selective agar base (modified CCDA-Preston) (Oxoid) for *C. jejuni* and SS agar (Oxoid) for *S. boydii*. The isolated bacterial species were identified by the Oxoid biochemical identification system.

To test the effect of atmospheric pressure cold plasma on the isolated bacterial species, suspensions of each individual bacterial species containing about  $10^7$  cells  $\text{ml}^{-1}$  in saline solution were prepared from colonies previously grown on the defined medium for 48 h. An aliquot (2 ml) of the bacterial suspension was transferred to a 2.5 ml capacity Eppendorf tube (Sigma) and exposed to cold plasma for 30, 60, 90, 120

and 150 s. The number of CFU in the bacterial suspension before and after cold plasma treatment was determined by standard plate counting as a reference method and using the spectrophotometric enumeration method (Shin *et al* 2007).

### 2.4. Endotoxins activity assay

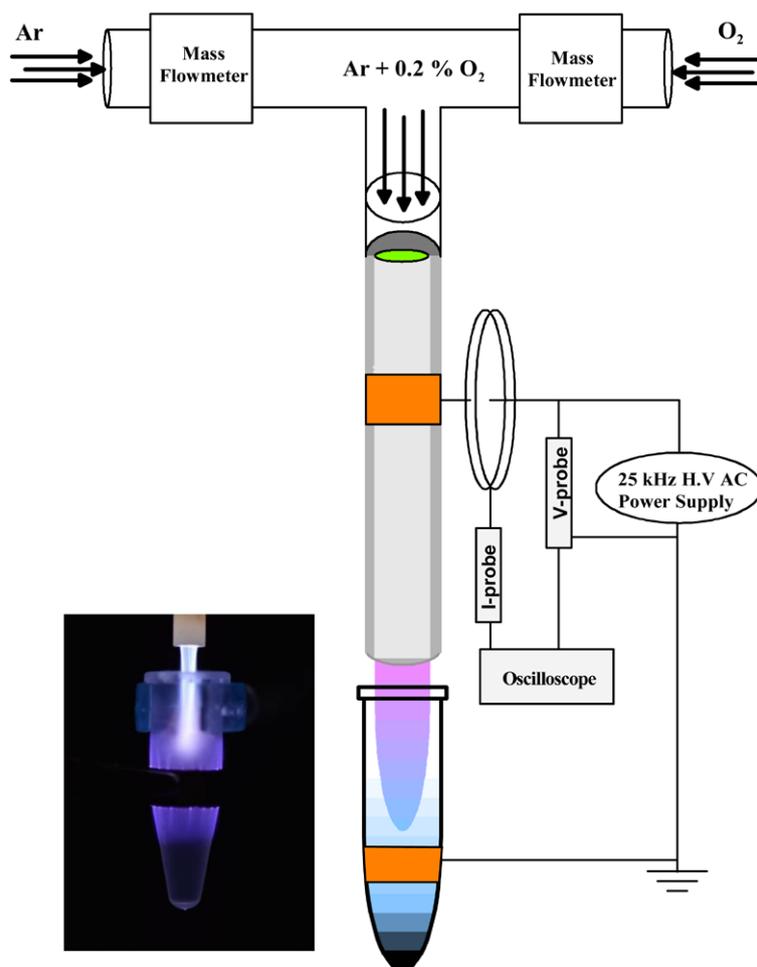
Endotoxin activity was analyzed according to the method previously described by Tsuji *et al* (1984) using chromogenic limulus (*Tachypleus tridentatus*) amoebocyte lysate (LAL) endpoint assay kits (Xiame Houshiji Ltd, China). The wastewater samples were diluted in pyrogen-free diluents based on the expected endotoxin concentrations. The diluted samples were mixed with LAL, incubated for 8 min at  $37^\circ\text{C}$ , mixed with the chromogenic substrate, and incubated for an additional 6 min. The reaction was stopped with 20% acetic acid. The absorbance of the developed yellow color was read at 405 nm in a microplate reader (Molecular Devices, Sunnyvale, CA, USA). The endotoxin activity was obtained from the standard curve using endotoxin standards from the *E. coli* O111:B4 strain.

### 2.5. Plasma jet setup

The atmospheric pressure cold plasma jet system consisted of a cylindrical alumina ( $\text{Al}_2\text{O}_3$ ) insulator tube with an outer diameter of 4.8 mm and 152 mm length. The tube had a capillary with hole diameter of 3 mm. A high voltage 25 kV AC sinusoidal wave, with  $\sim 25$  kHz frequency, was applied to a copper ring electrode surrounding the alumina tube. The high voltage electrode was 7 mm width and its lower end far from the jet nozzle at 20 mm. One ml of the sample was inserted into the capacity Eppendorf tube surrounded by a second identical copper ring acting as ground electrode as shown in figure 1. The bottom end of the ground electrode was located 22 mm away from the jet nozzle. A high voltage electric probe (Tektronix P6015) and Person current monitor probe (model 6585) were used to measure the voltage and current waveforms, respectively. The probes were connected to a 1 GHz,  $5 \text{ GS s}^{-1}$  digital phosphor oscilloscope (Tektronix DPO 4104B). In this work, the current and the high voltage waveform signals were measured at the high voltage electrode. The jet was formed by blowing argon gas through the alumina tube when a high voltage signal was applied between the two electrodes. Oxygen gas was mixed with argon at the rate of 0.2% from argon and the argon mass flow was fixed at 4.5 slm as shown in figure 1. The gas flow rates were measured using ALICAT MC-5SLPM-D/5 and ALICAT MC-20SLPM-D/5 mass flow meters for oxygen and argon, respectively. The temperature of the treated water sample was measured immediately after treatment using a thermocouple probe.

### 2.6. Power consumption

A Lissajous figure, charge–voltage (Q–V) characteristics, was estimated using a capacitance C means of 15 nF as explained previously by Eliasson and Kogelschatz (1991). The energy dissipated during one period of the applied voltage corresponds



**Figure 1.** Atmospheric pressure argon plasma experimental setup (right) and a picture of the plasma interacting with a treated wastewater sample (left).

to the enclosed area in the Lissajous figure according to the following equation:

$$P = \frac{1}{T} \int_0^T u(t)i(t) dt.$$

The power consumption is estimated by multiplying the consumed energy by the operating frequency. The power consumption was estimated at 12.5 kV of peak to peak applied voltage and 27 kHz of plasma operating parameters.

### 2.7. Emission spectra measurements

The emission spectra of the generated atmospheric pressure Ar/O<sub>2</sub> plasma jet in air were investigated by optical emission spectroscopy using an Acton SP-2356 imaging spectrograph. This system is a 0.5 m focal length spectrograph that has three gratings: 3600 gr mm<sup>-1</sup>, 1800 gr mm<sup>-1</sup>, and 150 gr mm<sup>-1</sup> blazed at 240 nm, 500 nm, and 500 nm, respectively. The spectrograph is equipped with a built-in photomultiplier detector with high sensitivity ranging from 190 to 900 nm (model ARC-P2, Princeton instrument). A UV-VIS optical fiber bundle was used to couple the spectra to the spectrograph. Through this work, the fiber bundle was oriented perpendicular to the long axis of the jet far from the nozzle at a distance of 2 mm.

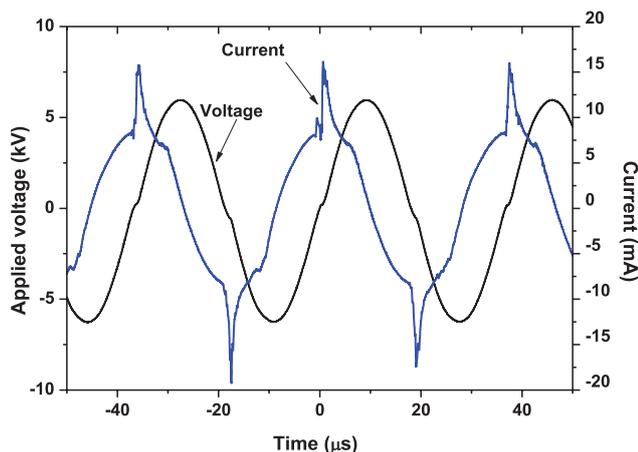
### 2.8. Statistical analysis

A statistical analysis (linear regression or ANOVA with statistical significance level fixed at  $p < 0.05$ ) was carried out using Microsoft Excel (Microsoft, Roselle, IL).

## 3. Results and discussion

### 3.1. Plasma electric characteristics

The current and voltage waveforms of 4.5 l min<sup>-1</sup> argon mixed with 0.009 l min<sup>-1</sup> (0.2%) oxygen are shown in figure 2. A sinusoidal waveform of 12.5 kV was applied to the high voltage electrode to produce about 36 mA peak to peak current. Wastewater from date palm fruits was used as a representative sample. The average current waveform revealed that the generated plasma consists of multi streamers, while fewer streamers, tending to be homogenous, were observed when pure argon was used as the operating gas. Moreover, an applied voltage instability was noticed for all the wastewater samples, particularly in the case of that derived from the blackberry processing industry (Tang *et al* 2010). The waveforms illustrate that the discharges ignite at ~4 kV for the positive and the negative half cycle of the applied voltage waveforms.



**Figure 2.** Voltage–current waveforms of an atmospheric pressure plasma of argon mixed with 0.2% oxygen interacting with wastewater derived from date palm fruit processing as a representative sample.

**Table 1.** Consumed power and energy for different wastewater samples necessary to achieve elimination or maximum reduction in bacterial count<sup>a</sup>.

| Source of wastewater | Consumed power (W) |                          | Energy density (kWh l <sup>-1</sup> ) |
|----------------------|--------------------|--------------------------|---------------------------------------|
|                      | Ar                 | Ar + 0.2% O <sub>2</sub> |                                       |
| Date palm            | 9.04 ± 1.03        | 9.02 ± 1.04              | 0.15                                  |
| Beetroot             | 9.13 ± 1.19        | 8.60 ± 1.03              | 0.21                                  |
| Blackberry           | 16.77 ± 2.55       | 15.95 ± 1.88             | 0.40                                  |
| Tomato               | 9.44 ± 1.70        | 8.14 ± 1.07              | 0.17                                  |

<sup>a</sup> The consumed energy was calculated by multiplying the consumed power by the time required for elimination or maximum reduction in bacterial count, which is different from sample to sample. Then the energy per unit volume was calculated by dividing the consumed energy by the volume of the treated wastewater sample (2 ml).

### 3.2. Plasma power consumption

The discharge consumed power in the case of pure argon plasma used in treating different wastewater samples was generally higher than the corresponding values in the case of argon mixed with 0.2% oxygen plasma (table 1). The average consumed power for the plasma interacting with different investigated samples ranged from 8.14 W (figure 3(a)) in the case of wastewater from tomato processing treated with argon mixed with 0.2% oxygen plasma, to 16.77 W (figure 3(b)) in the case of wastewater from blackberry processing treated with pure argon plasma.

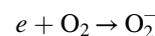
The consumed power is higher than that for a similar plasma in air without water for a discharge frequency of 20 kHz under a discharge of 8 kV in pure Ar plasma (Ying *et al* 2013). This higher value for the power consumption in this work is due to the use of water inside the discharge zone and the higher operating voltage (12.5 kV). As a comparison, Leverenz *et al* (2006) indicated that the energy usage of ultraviolet light used for the disinfection of wastewater effluent with biological oxygen demands and total suspended solids less than 30 mg l<sup>-1</sup> at a flow rate up to 3 gal min<sup>-1</sup> was 35 W, whereas it was 23 W for the disinfection of drinking water.

### 3.3. Plasma emission spectra

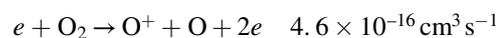
The emission spectra of pure argon and argon mixed with 0.2% O<sub>2</sub> plasma were compared in the ranges 200–450 nm and 450–850 nm (figure 4). Very strong argon lines in the range 650–850 nm and in the N<sub>2</sub> second positive band system (C<sup>3</sup>Π<sub>u</sub> → B<sup>3</sup>Π<sub>u</sub>) in both UV and visible ranges were recorded. The N<sub>2</sub> band is present due to the interaction of the argon plasma jet with air, since it operates at atmospheric pressure in open air and the spectra is measured 2 mm below the jet nozzle. The emission spectra indicate the generation of reactive radicals, mainly OH (A<sup>2</sup>Σ<sup>+</sup> → X<sup>2</sup>Π<sub>3/2</sub>), in the range 305–312 nm, and O radicals at 777.4 nm.

The most expected radicals that participated in microbial decontamination are OH (A<sup>2</sup>Σ → X<sup>2</sup>Π) and O (777.4 nm). The spectra of OH and O radicals differ according to the nature of the wastewater and the quantitative ratio of the operating gases, whether pure argon or argon mixed with oxygen. In preliminary tests, the optimum concentration of oxygen mixed with argon to achieve maximum decontamination was 0.2%. Figure 5 shows that the O (777.4 nm) emission spectrum is highly promoted on using argon mixed with 0.2% oxygen. This promotion enhances the plasma sterilization efficiency for wastewater treated samples. The maximum increase in the O (777.4 nm) spectrum was 608.4% recorded when plasma interacted with beetroot processing wastewater as compared to a relatively smaller increase in the case of blackberry processing wastewater (75.4%) as shown in table 2. On the other hand, the relative intensity of OH radicals generated from argon mixed with 0.2% oxygen showed little increase, ranging from 24.1% to 80.8% in the case of plasma interacting with wastewater of the different tested samples, except blackberry, which induced a reduction in OH radicals (–21.2%). The variation in the O and OH radical intensities are probably responsible for the differential efficacy of plasma as a decontamination agent.

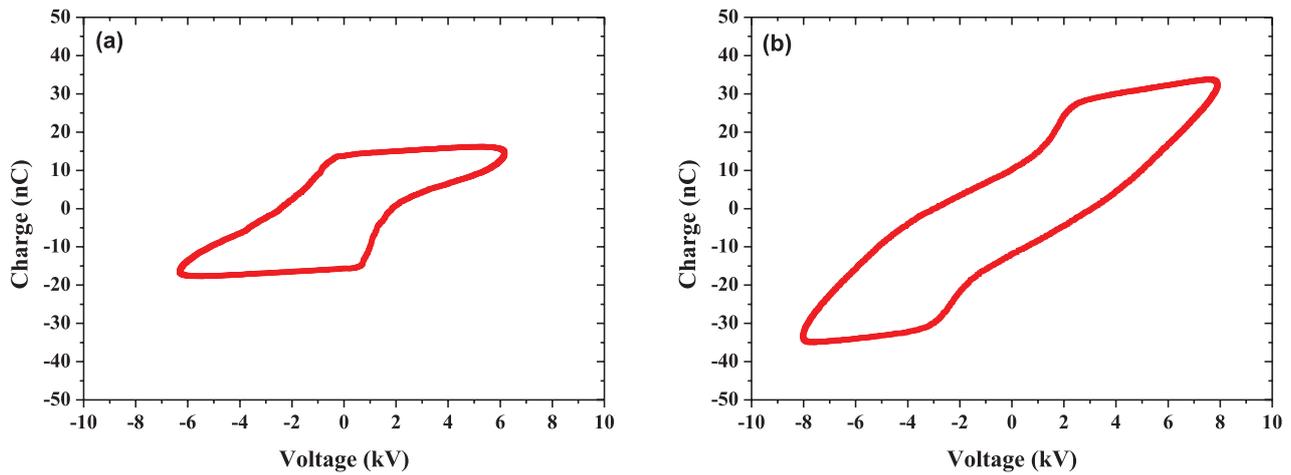
The generated plasma free electron mean energy in the discharge decreased due to oxygen addition, which is mainly related to the electron attachment. Adding oxygen will capture a lot of free electrons, forming negative oxygen molecules (O<sub>2</sub><sup>-</sup>); this process is enhanced when the plasma is operated in humid air (Penetrante *et al* 1997, Wang *et al* 2004) as follows:



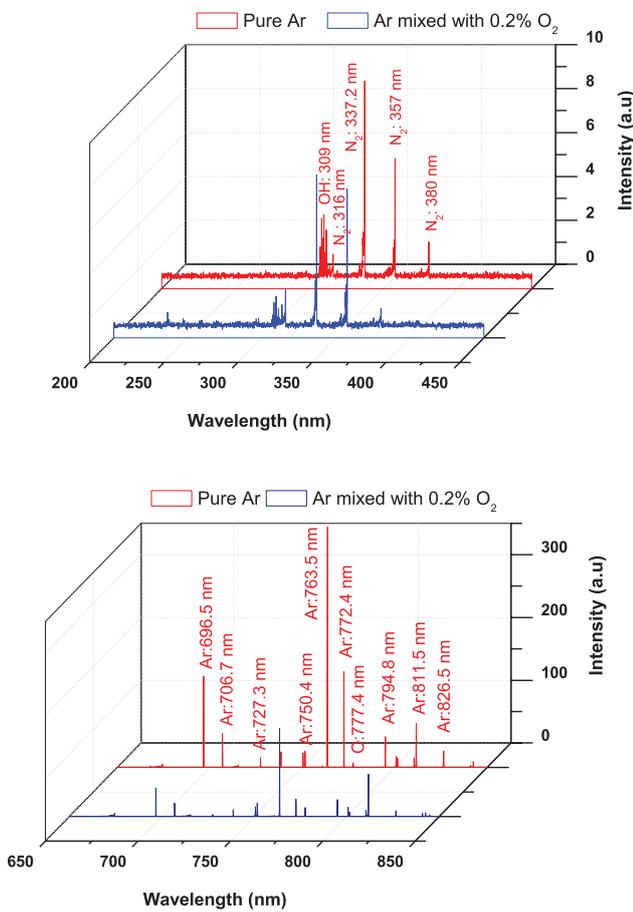
The oxygen molecule dissociates and a lot of [O] and [O] (1D) are generated and produce OH on interaction with H<sub>2</sub>O in humid air as follows (Yang *et al* 2008):



However, OH radicals may be also produced through the dissociative reactions of water cluster ions (Sigmund 1984).



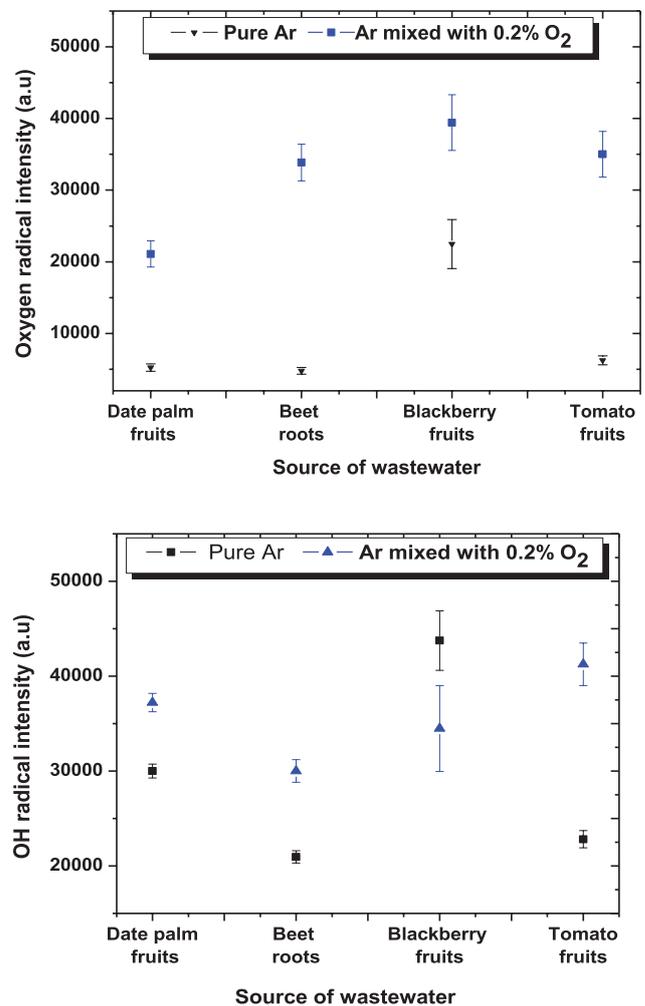
**Figure 3.** Examples of Lissajous figures, charge–voltage (Q–V) characteristics in the case of (a) wastewater from tomato processing treated with argon mixed with 0.2% oxygen plasma, 8.14 W, (b) wastewater from blackberry processing treated with pure argon plasma, 16.77 W.



**Figure 4.** A comparison between emission spectra of atmospheric pressure plasma jets of argon and argon mixed with 0.2% oxygen interacting with blackberry processing wastewater in the ranges 250–450 nm (above) and 650–850 nm (below).

The above reactions show that the free electrons are depleted on mixing argon with 0.2% O<sub>2</sub>. The process leads to an increase in the plasma electric instability and the tendency to form streamer plasma.

On the interaction of plasma with water, many chemical reactions may occur. These reactions could generate some



**Figure 5.** Comparison of the emission spectra of O radicals, 777.4 nm (above), and OH radicals, central band 309 nm (below), from argon and argon mixed with 0.2% oxygen plasma interacting with different fruit and vegetable processing wastewaters.

chemical species inside the water such as nitrite and nitrate ions (NO<sub>2</sub><sup>-</sup>, NO<sub>3</sub><sup>-</sup>), hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>), ozone (O<sub>3</sub>) and hydrogen molecule H<sub>2</sub> in addition to the formation of some

**Table 2.** Percentage change in OH and O (777.4 nm) radical emission spectra on applying an argon mixed with 0.2% oxygen plasma in relation to pure argon plasma on treating fruit and vegetable processing wastewater samples.

| Source of wastewater | Percentage change |       |
|----------------------|-------------------|-------|
|                      | O                 | OH    |
| Blackberry           | +75.4             | -21.2 |
| Date palm            | +302.5            | +24.1 |
| Tomato               | +460.5            | +80.8 |
| Beetroot             | +608.4            | +43.3 |

radicals (OH•, H•, O•) (Ruma *et al* 2014). These reactive chemical species possibly participate in the inactivation of the microorganisms existing in the water (Oehmigen *et al* 2010, 2011). Moreover, the reactive nitrogen species could be formed on the interaction of atmospheric pressure plasma with water, depending on the physical processes induced by the plasma. These processes are controlled by many variable parameters such as the water conductivity, plasma electron density, temperature and the operating gas. The main reactions involved in the production of free radicals and their termination on the reaction of the plasma with water have been summarized previously (Malik *et al* 2001, 2010). The thermocouple measurements show an increase in the treated sample temperature above its original room temperature (20 °C) by 3–5 °C, which indicates that the temperature does not have a role in sterilization and the decontamination is mostly related to the free radicals. The maximum increase in sample temperature was recorded in the case of wastewater from blackberry processing.

### 3.4. Plasma treatment

**3.4.1. Physical and chemical characteristics.** The wastewater produced from blackberry processing had higher carbohydrate and protein content and COD, while that produced from tomato processing showed the lowest values of these materials (table 3). The variation in wastewater characteristics is probably due to the type of plant material used, manufacture, nature of product, different additives, flavors, and the processes involved in the production including washing, peeling, blanching, slicing (or dicing), canning and retorting. The wastewater from blackberry processing has the highest antioxidant capacity (table 4), which is largely attributed to the presence of anthocyanins, phenolics and other flavonoids (Wang and Lin 2000, Siriwoharn *et al* 2004, Ortiz *et al* 2013). In their investigation on berries as natural antioxidants, Huang *et al* (2012) showed that blackberry fruits contain a range of phenolic acids and flavonoids (flavone, flavonols, flavanols, anthocyanidins) that have a remarkably high scavenging activity toward chemically generated radicals.

A significant reduction in total solids, total organic carbon and total nitrogen content, and pH values of the investigated wastewater samples was observed after 180s exposure to the cold plasma (table 3). The lowest reduction was observed in the case of wastewater from blackberry processing where the pH dropped to only 5.5 compared to 5.9 recorded before

treatment. It is possible that the reaction of the generated reactive species, including N, O, OH and ozone, with the wastewater components, such as organic acids and proteins, are mainly responsible for the change in pH to be more acidic (Fernandez and Thompson 2012). The change of the media pH towards the acidic range enhances plasma bactericidal efficacy (Liu *et al* 2010).

The efficiency of cold atmospheric argon plasma in reducing COD in the test samples after 180s exposure time reached 93.3 and 84.4% in the case of wastewater from date palm and tomato processing, respectively. The lowest reduction in COD was recorded in the case of wastewater from blackberry processing (57.5%). The removal of COD seems to depend on the nature of the wastewater. Spectroscopy shows that the plasma content includes chemical reactive species such as O and OH radicals as well as excited molecular nitrogen and argon. The hydroxyl radicals have been shown by several investigators to aggressively attack a broad range of organic contaminants and micro pollutants (Aleboye *et al* 2008, Wu and Linden 2008, Yuan *et al* 2009, Olmez-Hanci *et al* 2011). Grundmann *et al* (2007) reported that the produced oxidizing reactive species can decompose pollutant molecules, organic particulate matter or soot, resulting in a reduction of COD. One of the main limitations of this process, however, is that the treatment efficiency is significantly influenced by the water quality, such as the presence of natural organic matter, and/or alkalinity (He *et al* 2012).

The relatively high reduction in COD in the case of date palm and tomato processing wastewater may be due to the relatively lower antioxidant capacity that results in a longer life span of the produced reactive species as compared with the depletion of the reactive species resulting from the higher antioxidants in the case of wastewater from blackberry processing. Wang and Jiao (2000) stated that the blackberry has the highest scavenger capacity against superoxide radicals, hydrogen peroxide, hydroxyl radicals and singlet oxygen. Moreover, spectral analysis shows instability in the produced plasma in the wastewater of blackberry processing, probably due to the high antioxidant power that decreases the strength of the reactive species. The reactions between the oxygen ions and chemicals present in the wastewater will generally produce inert gases and/or precipitates with little or no negative impact.

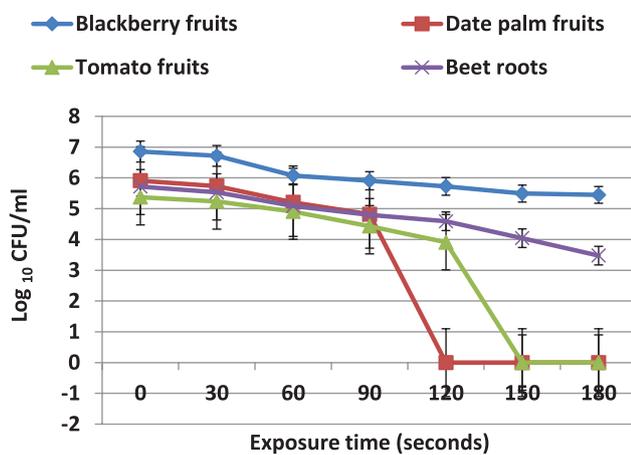
**3.4.2. Bacterial count.** The wastewater from fruit and vegetable processing operations varied in bacterial load according to the processed material (figure 6). Wastewater from blackberry processing had the highest bacterial population at 6.85 log<sub>10</sub> CFU ml<sup>-1</sup>, followed by that from date palm, beetroot and tomato processing, reaching 5.91, 5.72 and 5.37 log<sub>10</sub> CFU ml<sup>-1</sup>, respectively. The most common bacterial species were identified as *E. coli*, *P. aeruginosa*, *E. faecalis*, *C. jejuni* and *S. boydii* (figure 7). Initially, fruits and vegetables are commonly exposed to microbial contamination through contact with soil composted with organic manure, water and dust, and through handling at harvest or during post-harvest processing. Contamination may also be due to storage conditions and how long the wastewater was kept before it

**Table 3.** Physical and chemical characteristics of fruit and vegetable processing wastewater before (BT) and after (AT) atmospheric pressure plasma jet treatment of argon mixed with 0.2% oxygen applied for 180s.

| Source of wastewater | pH        |           | Total solids (TS) |          | Total organic carbon (mg l <sup>-1</sup> ) |          | Total nitrogen (mg l <sup>-1</sup> ) |        | COD (mg l <sup>-1</sup> )m |          |
|----------------------|-----------|-----------|-------------------|----------|--|----------|--------------------------------------|--------|----------------------------|----------|
|                      | BT        | AT        | BT                | AT       | BT   | AT       | BT                                   | AT     | BT                         | AT       |
| Blackberry           | 5.9 ± 0.2 | 5.5 ± 0.2 | 840 ± 26          | 629 ± 22 | 260 ± 12                                   | 155 ± 11 | 92 ± 8                               | 58 ± 6 | 930 ± 28                   | 395 ± 13 |
| Date palm            | 6.1 ± 0.3 | 5.3 ± 0.3 | 471 ± 19          | 261 ± 14 | 110 ± 8                                    | 31 ± 7   | 37 ± 5                               | 12 ± 3 | 410 ± 15                   | 27 ± 4   |
| Tomato               | 5.3 ± 0.3 | 4.6 ± 0.2 | 322 ± 15          | 143 ± 14 | 88 ± 7                                     | 26 ± 5   | 21 ± 4                               | 9 ± 2  | 294 ± 16                   | 45 ± 6   |
| Beetroot             | 6.2 ± 0.4 | 5.6 ± 0.3 | 630 ± 18          | 312 ± 16 | 170 ± 10                                   | 44 ± 8   | 43 ± 5                               | 19 ± 4 | 501 ± 19                   | 141 ± 10 |

**Table 4.** Ferric reducing antioxidant power (mmol Fe<sup>2+</sup> l<sup>-1</sup>) of fruit and vegetable processing wastewater samples.

| Source of wastewater | Antioxidant power (mmol Fe <sup>2+</sup> l <sup>-1</sup> ) |
|----------------------|--|
| Blackberry           | 14.82  |
| Date palm            | 1.03   |
| Tomato               | 3.34   |
| Beetroot             | 5.70   |



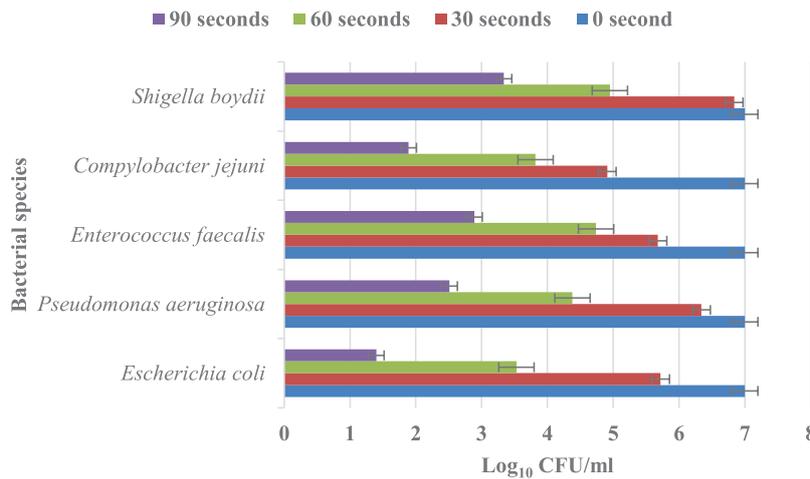
**Figure 6.** Effect of atmospheric pressure plasma of argon mixed with 0.2% oxygen on bacterial count (log<sub>10</sub> CFU ml<sup>-1</sup>) of fruit and vegetable processing wastewaters.

was obtained for sampling. The wastewater probably contains different fermentable sugars, a broad spectrum of waste soluble organic materials, and essential elements originating in processing. These nutrients facilitate colonization of bacterial isolates and most likely account for the high microbial counts. More significantly, bacteria on the produce may multiply over time depending on the storage conditions, especially those that are psychrotrophic (Abadias et al 2008, Montville and Matthews 2008).

A number of experiments were carried out with different configurations of the atmospheric pressure argon cold plasma to study the best processing conditions that achieve the highest microbial inactivation. The effect of the atmospheric cold plasma treatment on the reduction of total count of bacteria in different wastewater samples is shown in figure 6. Plasma treatment for 150s induced complete decontamination of the wastewater derived from tomato processing, while 120s was sufficient for decontamination of wastewater from date palm processing. Meanwhile, plasma treatment induced

a significant reduction in the bacterial count by 1.41 and 2.43 log<sub>10</sub> CFU ml<sup>-1</sup> in the case of wastewater from blackberry and beetroot processing, respectively. The degree of inactivation probably depends on the type of microorganism, the inactivation medium, number of cells, operating gas mixture, gas flow and physiological state of cells (Yu et al 2006, Song et al 2009). The difference in effectiveness of reactive species produced by the atmospheric plasma is mostly due to the variation in the antioxidant capacity of the wastewater. Critzer et al (2007) reported the ability of atmospheric plasma to reduce inoculated microbial populations on fresh produce surfaces. Fernandez et al (2013) stated that cold atmospheric gas plasma achieved 2.72, 1.76 and 0.94 log reductions of *Salmonella typhimurium* on lettuce, strawberry and potato, respectively, in 15 min of plasma treatment time.

Considering the effect of the cold plasma on isolated bacterial species in saline (figure 7), *E. coli* was the most susceptible bacterial species, and its count dropped to 1.4 log<sub>10</sub> CFU ml<sup>-1</sup> after 90s exposure to cold plasma, compared to 7.0 log<sub>10</sub> CFU ml<sup>-1</sup> for the initial untreated sample. *S. boydii*, on the other hand, had the minimum susceptibility, recording 3.34 log<sub>10</sub> CFU ml<sup>-1</sup> as compared to the same initial count. All isolated species failed to grow on extension of the exposure time to 120s. It is suggested that charged particles of plasma can inactivate bacterial cells through the rupture of the cytoplasmic membrane of bacterial cells, where the electrostatic force caused by charge accumulation on the outer surface of the cell membrane overcomes the tensile strength of the membrane, thereby causing its rupture (Mendis et al 2000, Laroussi et al 2003). Surowsky et al (2014) suggested that the load of *Citrobacter freundii* in apple juice was reduced by about 5 log cycles after plasma exposure of 480s using argon and 0.1% oxygen followed by a subsequent storage time of 24h. The results indicate that direct contact between bacterial cells and cold plasma is not necessary for achieving successful inactivation. They also showed that the plasma generated compounds in the liquid, such as H<sub>2</sub>O<sub>2</sub> and most likely hydroperoxy radicals, are particularly responsible for microbial inactivation. van Gils et al (2013) used a radio-frequency atmospheric pressure argon plasma jet to inactivate *P. aeruginosa* in solutions, and showed that HNO<sub>2</sub>, ONOO<sup>-</sup> and H<sub>2</sub>O<sub>2</sub> detected in the liquid phase were the responsible species for inactivation. The effectiveness of the cold plasma on bacterial species contained in saline appears more efficacious than that recorded for the bacteria occurring in wastewater, undoubtedly due to the absence of antioxidant. Surowsky et al (2014) stated that the reactive species scavenging role of the



**Figure 7.** Effect of atmospheric pressure plasma of argon mixed with 0.2% oxygen on count ( $\log_{10}$  CFU  $\text{ml}^{-1}$ ) of the common bacterial species isolated from fruit and vegetable processing wastewater.

**Table 5.** Occurrence of endotoxins in fruit and vegetable processing wastewater samples treated with atmospheric pressure cold plasma jet of argon mixed with 0.2% oxygen for 180s.

| Source of wastewater | Endotoxins ( $\text{ng ml}^{-1}$ ) |                 | % reduction |
|----------------------|------------------------------------|-----------------|-------------|
|                      | Before treatment                   | After treatment |             |
| Blackberry           | $530.5 \pm 32$                     | $239.6 \pm 12$  | 54.84       |
| Date palm            | $272.1 \pm 22$                     | $26.6 \pm 4$    | 90.22       |
| Tomato               | $136.6 \pm 14$                     | $33.0 \pm 6$    | 75.84       |
| Beetroot             | $359.0 \pm 26$                     | $108.3 \pm 11$  | 69.83       |

antioxidants in the apple juice might play an important role in the attenuation of reactive species. The percentage change in OH and O radicals of the emission spectra (table 2) shows that the role of antioxidant is mostly related to OH radicals rather than O radicals, where the OH radicals are negatively changed in wastewater from blackberry processing (-43.3%). Although the cold plasma differentially promoted the O radicals of all tested samples, wastewater from blackberry processing showed the lowest increase of these radicals, probably due to the high antioxidant level.

**3.4.3. Toxin inactivation.** The bacterial endotoxins varied in different test wastewater samples. Wastewater from blackberry processing contains the highest concentration ( $530.5 \text{ ng ml}^{-1}$ ) and that of tomato processing has the lowest concentration ( $136.6 \text{ ng ml}^{-1}$ ), as indicated in table 5. The occurrence of endotoxin in wastewater from fruit and vegetable processing is expected and reflects the occurrence of the high level of Gram negative bacteria including *E. coli*, *P. aeruginosa*, *C. jejuni* and *S. boydii*. The acceptable endotoxin level that the body can tolerate without adverse effect ranges from 0.5 to  $0.02 \text{ ng kg}^{-1}$  of body weight per hour (US pharmacopeial Convention Inc. 2004). The effect of endotoxins on *in vitro* cell growth and function depends on their levels and cell type. Endotoxin levels as low as  $1.0 \text{ ng ml}^{-1}$  reduces pregnancy success rates in *in vitro* fertilization (Snyman and Van der Merwe 1986), while a higher level of  $100 \text{ ng ml}^{-1}$  induces proliferation of monocytes (Mattern et al 1994).

It is supposed that a considerable endotoxin increase in the medium of the cold plasma treated samples results from accumulation of the released lipopolysaccharide components in the cell walls of destroyed Gram-negative bacteria. However, the treatment of wastewater samples with cold atmospheric argon plasma for 180s induced a reduction in the endotoxins ranging from 90.22% (in the case of date palm) to 54.84% (in case of blackberry). Extension of the exposure time to 360s induced complete elimination of endotoxins (unrecorded data). He et al (2012) indicated that cyanobacterial toxin cylindrospermopsin, at an initial concentration of  $1 \mu\text{M}$ , was significantly degraded from 75% to 100% by means of OH radicals and sulfate radicals using UV-254nm activation of  $\text{H}_2\text{O}_2$ , persulfate and peroxymonosulfate. They also stated that certain radical scavengers in tap water samples inhibited the destruction of cylindrospermopsin by the OH radicals. It is suggested that concentrations of antioxidants in different wastewaters significantly affect the oxidation potentialities of reactive species and consequently their efficiency in toxin breakdown.

#### 4. Conclusions

The wastewater produced from fruits or vegetables processing industries are a problem for both the environment and drainage systems, although it contains relatively small quantitative amounts of microorganisms of saprophytic and sometimes pathogenic nature. To allow for a more reliably safe environment, the recycling of this type of wastewater is essential and is easier to decontaminate than other hazardous wastewater. A cold atmospheric pressure plasma jet with argon or a mixture of argon and 0.2% oxygen as operating gases was generated and characterized. The jet was generated by blowing the gas through an alumina tube powered by a 25 kHz, 25 kV power supply. The plasma transfers from a single homogenous jet, when argon was used individually, to multi-streamers on its mixing with 0.2% oxygen. The generated plasma jet was employed to improve the physical and microbiological characteristics of wastewater derived from processing industries of blackberry, date palm, tomato and beetroot.

The plasma interacting with the wastewater samples was run in unstable mode and obviously penetrates the investigated samples. Regardless of the OH radicals that were reduced in case of treated blackberry fruits, the addition of 0.2% oxygen to the argon steadily induced an increase in the O and OH radical emission spectra in the case of wastewater from the all treated samples. The results indicate significant efficiencies of the cold plasma in improving the physical and chemical characteristics of the tested wastewater. The maximum COD reduction of wastewater samples was counted in the case of date palm (93.3%) and the minimum one was recorded for that of blackberry (57.5%). Although the bacterial load of wastewater from blackberry and beetroot processing was significantly reduced after 180s, the cold plasma was able to completely decontaminate the wastewater from date palm and tomato processing after 120 and 150 s, respectively. The effectiveness of the cold plasma on the isolated bacterial species *E. coli*, *P. aeruginosa*, *C. jejuni*, *S. boydii* and *E. faecalis* contained in saline appears more efficacious than that recorded for the bacteria occurring in wastewater, undoubtedly due to the absence of antioxidants. It is interesting to note that the variation in cold plasma effectiveness was related to the antioxidant activity of each type of wastewater. The higher the antioxidant power, the lower the efficacy of the cold plasma. The plasma efficiency, particularly on using a mixture of argon and 0.2% oxygen, as a decontamination agent was reduced by antioxidant constituents in wastewater that act as scavengers for the active free radicals produced by plasma and recorded by spectral data. The reduction of endotoxins in water ranged from 54.84, in the case of wastewater from blackberry processing, to 90.22 in the case of wastewater from date palm processing. This indicates the ability of cold plasma to degrade the bacterial toxins, and possibly eliminate them completely on extension of exposure time.

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